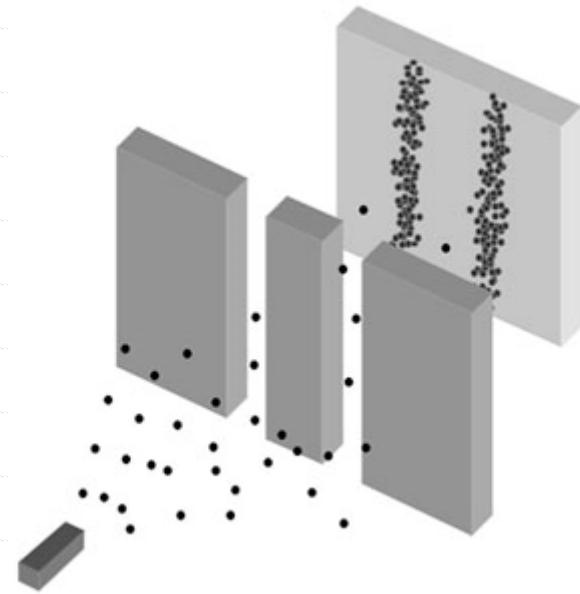


Chapter 29: Particles and Waves

Essential Concepts and Summary

Wave-Particle Duality

- ◆ Waves can exhibit particle-like characteristics, and particles can exhibit wave-like characteristics
- ◆ Can be seen through a version of Young's experiment, using a beam of electrons



Blackbody Radiation and Planck's Constant

- ◆ All bodies radiate E/M waves continually
- ◆ Assuming energy quantization, Planck obtained energy of these waves
- ◆ Suggests light consists of energy packets, not continuous waves

$$E = nhf$$

$$n = 0, 1, 2, 3, \dots$$

$$h = 6.63 \cdot 10^{-34} \text{ J} \cdot \text{s}$$

Photons and Photoelectric Effect

- ◆ Photoelectric effect: electrons emitted from metal surface when light shines on it.
- ◆ Because electrons ejected with aid of light, called photoelectrons.
- ◆ Einstein proposed energy of a photon given by relation between planck's constant and frequency, not light intensity
- ◆ Work function is energy necessary to eject electron

$$E = hf$$

$$hf = KE_{\max} + W_0$$

Momentum of Photon

- ◆ Compton effect: X-ray photon scattered from electron, with scattered photon having smaller frequency than incident photon.
- ◆ Using Compton photon and conservation of momentum, we can derive momentum of photon

$$\lambda' - \lambda = \frac{h}{mc} (1 - \cos \theta)$$

$$p = \frac{E}{c} = \frac{hf}{f\lambda} = \frac{h}{\lambda}$$

De Broglie Wavelength and Wave Nature of Matter

- ◆ De Broglie made proposal that wavelength of particle is governed by the same relation as what applies to a photon.
- ◆ Waves of particles are waves of probabilities, whose magnitudes give indication of probability that particle will be found at that point

$$\lambda = \frac{h}{p}$$

Heisenberg Uncertainty Principle

- ◆ There are limits on the accuracy with which momentum and position can be determined.

$$(\Delta p_y)(\Delta y) \geq \frac{h}{4\pi}$$

- ◆ These are fundamental limits, not measuring errors.

$$(\Delta E)(\Delta t) \geq \frac{h}{4\pi}$$

- ◆ There are limits between momentum and position; between energy and time

Summary of Equations

$$E = hf$$

$$hf = KE_{\max} + W_0$$

$$\lambda' - \lambda = \frac{h}{mc}(1 - \cos \theta)$$

$$p = \frac{h}{\lambda}$$

$$\lambda = \frac{h}{p}$$

$$(\Delta p_y)(\Delta y) \geq \frac{h}{4\pi}$$

$$(\Delta E)(\Delta t) \geq \frac{h}{4\pi}$$

References

- ◆ <http://www.blacklightpower.com/theory/DoubleSlit.shtml>